



ULTIMATE TEST SERIES NEET -2020

XII TEST-3 SOLUTION

Test Date :21-03-2020

[PHYSICS]

1. $I_B = \frac{Mr^2}{2} + \left[\frac{Mr^2}{2} + M(2r)^2 \right] = 5Mr^2$

2. Net force on motor will be
 $F_m = [920 + 68(10)]g + 6000$
 $= 22000 \text{ N}$
So, required power for motor
 $P_m = \vec{F}_m \cdot \vec{v}$
 $= 22000 \times 3$
 $= 66000 \text{ watt}$

3. $I = Mr^2 = (2\pi r \lambda) R^2, I \propto R^3 \quad \frac{1}{8} = \left[\frac{R}{nR} \right]^3, n = 2$

4. $x = \frac{0 \times \pi(28)^2 - 7 \times \pi(21)^2}{\pi(28)^2 - \pi(21)^2}$

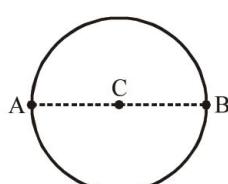
$x = -\frac{7\pi(21)^2}{\pi \times 7 \times 49} = -9 \text{ cm}$
distance from origin = 9 cm

5. $\frac{I_2}{I_1} = \frac{m_2 r_2^2}{m_1 r_1^2} = \frac{(A \times 2\pi r_2) \rho r_2^2}{(A \times 2\pi r_1) \rho r_1^2} = \frac{r_2^3}{r_1^3}$

$\frac{r_2}{r_1} = (4)^{1/3}$

6. $\frac{\omega_A}{\omega_C} = \frac{\frac{V_A}{r_A}}{\frac{V_C}{r_C}} = \frac{r_C}{r_A}$

$= \frac{a}{2a} = \frac{1}{2}$



7. $I = \frac{2}{5} MR^2$

$I' = \frac{2}{5} MR^2 + \frac{2}{5} MR^2 + \frac{7}{5} MR^2 + \frac{7}{5} MR^2$

$I' = \frac{18}{5} mR^2 = 9I$

8. $mgh = \frac{1}{2} mv^2 + \frac{1}{2} Iw^2$

$mgh = \frac{1}{2} mv^2 + \frac{1}{2} \frac{MR^2}{2} \times \frac{v^2}{R^2}$

$v = \sqrt{\frac{4mgh}{2m+M}}$

9. $(2 \times 3)\hat{i} + (1 \times 4)(-\hat{i}) = (2+1)\vec{v}$

$2\hat{i} = 3\vec{v}$

$\vec{v} = \frac{2}{3}\hat{i} \text{ m/s}$

10. $\vec{\tau} = \frac{d\vec{J}}{dt} = \frac{d}{dt} (\vec{r} \times \vec{p}) = m \frac{d}{dt} (\vec{r} \times \vec{v})$

$\vec{\tau} = [16 t]\hat{k}$

at $t = 2 \quad \vec{\tau}_{t=2} = (16 \times 2)\hat{k} = 32\hat{k} \text{ N-m}$

11. $\vec{r}_{cm} = \frac{m_1 \vec{r}_1 + m_2 \vec{r}_2 + m_3 \vec{r}_3}{m_1 + m_2 + m_3}$

12. $P = \tau \cdot \omega$

$$P = I \alpha \cdot \omega = I \left(\omega \frac{d\omega}{d\theta} \right) \cdot \omega$$

$$\omega^2 d\omega = \frac{P}{I} d\theta$$

$$\omega \propto \theta^{1/3}$$

$$\omega \propto (n)^{1/3}$$

13. B

14. $0.3X = 0.7(1.4 - X)$
 $x = 0.9 \text{ m}$

15. C

16. Moment of inertia of solid sphere of mass M and radius R about an axis passing through the centre of mass is: $I = \frac{2}{5}MR^2$. Let the radius of the disc is r.

Moment of inertia of circular disc of radius r and mass M about an axis passing through the centre of mass and perpendicular to its plane $= \frac{1}{2}Mr^2$.

Using theorem of parallel axes, moment of inertia of disc about its edge is:

$$I' = \frac{1}{2}Mr^2 + Mr^2 = \frac{3}{2}Mr^2$$

Given : $I = I'$

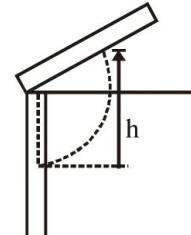
$$\text{or } \frac{2}{5}MR^2 = \frac{3}{2}Mr^2$$

$$\text{or } r^2 = \frac{4}{15}R^2$$

$$\text{or } r = \frac{2R}{\sqrt{15}}$$

17. $TE_i = TE_f$

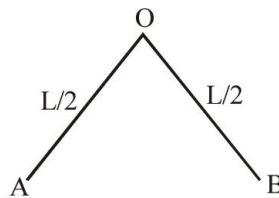
$$\frac{1}{2}I\omega^2 = mgh$$



$$\frac{1}{2} \times \frac{1}{3}m\ell^2\omega^2 = mgh$$

$$\text{or } h = \frac{1}{6} \frac{\ell^2\omega^2}{g}$$

18.



Total mass = M, total length = L

Moment of inertia of OA = OB about Q

$$= MI_{\text{total}} = 2 \times \left(\frac{M}{2} \right) \times \left(\frac{L}{2} \right)^2 \cdot \frac{1}{3} = \frac{ML^2}{12}.$$

19. For pure translatory motion of object, the force should act at the centre of mass.

$$Y_{CM} = \frac{m \times 2\ell + 2m \times \ell}{3m} = \frac{4\ell}{3}.$$

20. $\frac{1}{2}MV^2 = \frac{1}{2}KL^2$

$$V^2 = \frac{K}{M}L^2 \Rightarrow V = \sqrt{\frac{K}{M}}L \Rightarrow P = M\sqrt{\frac{K}{M}}L = \sqrt{MK}L$$

21. $\vec{r}_{cm} = \frac{m_1 \vec{r}_1 + m_2 \vec{r}_2}{m_1 + m_2}$

22. Required fraction

$$= \frac{K_R}{K_R + K_T} = \frac{\frac{1}{2} I \omega^2}{\frac{1}{2} I \omega^2 + \frac{1}{2} M v^2}$$

$$= \frac{\frac{1}{2} M R^2 \omega^2}{\frac{1}{2} M R^2 \omega^2 + \frac{1}{2} M v^2}$$

$$= \frac{M R^2 (v^2 / R^2)}{M R^2 (v^2 / R^2) + M v^2}$$

$$= \frac{M v^2}{M v^2 + M v^2} = \frac{1}{2}$$

23. $\vec{v}_{cm} = \frac{m_1 \vec{v}_1 + m_2 \vec{v}_2}{m_1 + m_2}$

24. From given graphs :-

$$F_x = \frac{3}{4}x + 10, F_y = -\frac{4}{3}y + 20, F_z = \frac{4}{3}z - 16$$

$$W = \int \vec{F} \cdot d\vec{r} = \int_0^8 F_x dx + \int_5^{20} F_y dy + \int_{12}^0 F_z dz$$

$$W = 104 + 50 + 96 = 250 \text{ J}$$

25. D

26. $W = \frac{1}{2} \times 5 \times 10^3 [(10 \times 10^{-2})^2 - (5 \times 10^{-2})^2]$

$$W = 18.75 \text{ N-m}$$

27. B

28. $\Delta u = n C_v \Delta T = n \frac{R}{\gamma - 1} \Delta T = \frac{p \Delta V}{\gamma - 1} = \frac{p V}{\gamma - 1}$

29. $\therefore F \propto V$

$$\therefore P = V^2 \frac{dm}{dt} \quad [\sqrt{P} \propto V]$$

30. C

31. A

32. A

33. MI of disc about diametric axis will be minimum.

34. Sphere compresses the spring until its all K.E. is converted to P.E. of spring

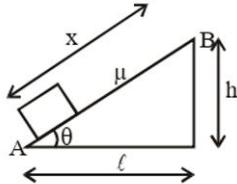
$$\frac{1}{2} M V^2 \left(1 + \frac{K^2}{r^2} \right) = \frac{1}{2} K x^2$$

35. A

36. B

37. C

38.



$$mg \sin \theta + \mu mg \cos \theta)x$$

$$Mg \left(\frac{h}{x} + u \frac{1}{x} \right) . x$$

$$Mg (h + \mu l)$$

39. $\Delta KE = \frac{1}{2} \frac{m_1 m_2}{m_1 + m_2} (1 - e^2) (u_1 - u_2)^2$

$$\Delta KE = \frac{1}{2} \times \frac{m}{2} (1 - e^2) u^2 = \frac{1}{4} \times \left(\frac{1}{2} m u^2 \right)$$

$$\Rightarrow \frac{1 - e^2}{2} = \frac{1}{4}$$

$$\Rightarrow e^2 = 1/2 \Rightarrow e = \frac{1}{\sqrt{2}}$$

40. $\frac{4m_1 m_2}{(m_1 + m_2)^2}$

41. $W = \frac{m(g \sin 30^\circ)l}{2n^2}$

42. Power = $F_{ext} \cdot v$
 $= (m_2 - m_1)g \cdot v$

43. $I_{net} = I_{disc} - I_{removed}$

$$= \frac{1}{2} (9M)R^2 - \frac{1}{2} M\left(\frac{R}{3}\right)^2 = \frac{40}{9} MR^2$$

44. D

45. Here $\frac{dv}{dt} = \text{constant} = a$ (say)

Use $v^2 = u^2 + 2as$ where

$s = 2 \times 2\pi r = 80$ m, $u = 0$, $v = 80$ m/s

[CHEMISTRY]

46. (B)

47. (A)

48. (D)

49. A

50. D

51. D

52. Given $\text{CuSO}_{4(s)} + \text{aq} \rightarrow \text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ (aq)

$$\Delta H = -15.9 \text{ kcal} \quad \dots(1)$$

$\text{CuSO}_{4(s)} \cdot 5\text{H}_2\text{O} + \text{aq} \rightarrow \text{CuSO}_4 + 5\text{H}_2\text{O}$ (aq)

$$\Delta H = +2.8 \text{ kcal} \quad \dots(2)$$

Subtracting equation (2) from equation (1)

$$\begin{aligned} \text{CuSO}_{4(s)} + 5\text{H}_2\text{O} &\rightarrow \text{CuSO}_4 \cdot 5\text{H}_2\text{O}_{(s)} \\ \Delta H &= -15.9 - 2.8 \\ &= -18.7 \text{ kcal} \end{aligned}$$

53. D

54. B

55. $\text{Ag}^+ + 2\text{NH}_3 \rightarrow [\text{Ag}(\text{NH}_3)_2]^+$; $k' = ?$

$$k' = k_1 \times k_2$$

56. B

57. $\text{HA} + \text{OH}^- \rightleftharpoons \text{A}^- + \text{H}_2\text{O}; k' = ?$

$$k' = \frac{1}{k_h} = \frac{1}{k_w/k_a} = \frac{k_a}{k_w} = \frac{10^{-4}}{10^{-14}} = 10^{10}$$

58. A

59. C

60.

$$S^1 = \frac{K_{sp}}{2C}$$

$$C = \frac{10\text{g}}{111\text{g/mol} \times 1\text{L}}$$

61. Option 4th is of weak base remaining all are salts of SAWB which have pH less than seven

62. B

63. D

$$K^1 = \frac{1}{\sqrt{K}}$$

65. $\Delta G = \Delta G^\circ + 2.303 RT \log_{10} Q$

66. C

67. Molarity (M) = $\frac{\text{wt}}{\text{mol.wt.}} \times \frac{1000}{\text{vol (ml)}}$

$$= \frac{25.3}{106} \times \frac{1000}{250}$$

= .955 mol/L of Na_2CO_3

and $\text{Na}_2\text{CO}_3 \longrightarrow 2\text{Na}^+ + \text{CO}_3^{2-}$

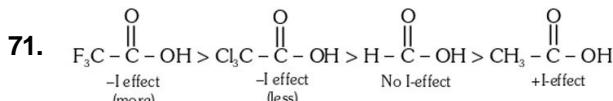
therefor $[\text{Na}^+] = 2 \times 0.955 = 1.910 \text{ M}$

$$[\text{CO}_3^{2-}] = 0.955 \text{ M}$$

68. B

69. A

70. C



72. Down the group in Gr -16 hydrides

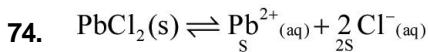
M-H bond length increases (due to increases in size)

Hence acidic nature increases

Hence $K_a \uparrow$ while $pK_a \downarrow$

73. Krichoff's equation

$$\Delta C_p = \frac{\Delta H_2 - \Delta H_1}{T_2 - T_1}$$



$$K_{sp} = 4s^3$$

$$1 \times 10^{-6} = 4s^3$$

$$s^3 = \frac{1}{4} \times 10^{-6}$$

$$= \left(\frac{1}{4}\right)^{\frac{1}{3}} \times 10^{-2}$$

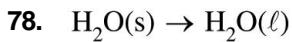
$$= 0.63 \times 10^{-2} = 6.3 \times 10^{-3}$$

75. $\int \Delta S = \frac{q}{T} = \int \frac{nC_p dT}{T}$

$$\Delta S = nC_p \ln \frac{T_2}{T_1}$$

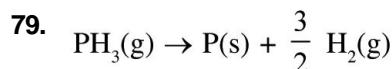
76. $pH = pK_a + \log \frac{[X^-]}{[HX]}$

77. $\Delta n = \oplus ve$
 $P \uparrow$ backward shifting



$$\Delta S = \frac{\Delta H}{T}$$

$$\Delta S = S_{\text{product}} - S_{\text{reactant}}$$



100 mL	0 mL
0 mL	150 mL

$$\Delta V = 150 - 100 = 50 \text{ mL}$$

80. $w = -2.303 nRT \log_{10} \frac{V_2}{V_1}$

81. $\Delta U = q + w$
 $w = -P_{\text{ext}} \cdot \Delta V$

82. $\Delta H - T\Delta S = 0$

83. $T \uparrow$ viscosity \downarrow

84. $NV = (N_1 V_1)_{\text{base}} - (N_2 V_2)_{\text{acid}}$

85. $pH = 7 + \frac{1}{2} pK_a - \frac{1}{2} pK_b$

86. A

87. C

88. B

89. C

90. $K_c = [H_2O]^2$. Solid phases are not to be reported.